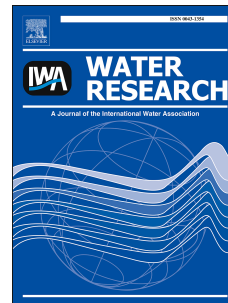


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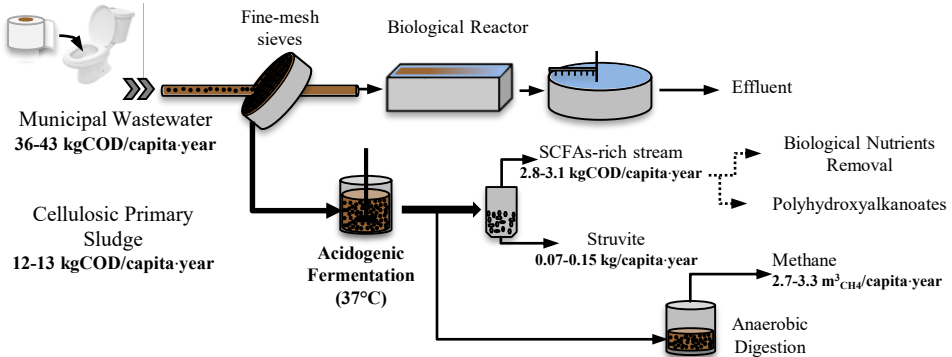
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1 **Biorefinery of cellulosic primary sludge towards targeted Short Chain Fatty Acids,**
2 **phosphorus and methane recovery**

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10

11 **Abstract**

12 Cellulose from used toilet paper is a major untapped resource embedded in municipal
13 wastewater which recovery and valorisation to valuable products can be optimized. Cellulosic
14 primary sludge (CPS) can be separated by upstream dynamic sieving and anaerobically
15 digested to recover methane as much as 4.02 m³/capita·year. On the other hand, optimal
16 acidogenic fermenting conditions of CPS allows the production of targeted short-chain fatty
17 acids (SCFAs) as much as 2.92 kg COD/capita·year. Here propionate content can be more
18 than 30% and can optimize the enhanced biological phosphorus removal (EBPR) processes or
19 the higher valuable co-polymer of polyhydroxyalkanoates (PHAs). In this work, first a full set
20 of batch assays were used at three different temperatures (37, 55 and 70 °C) and three
21 different initial pH (8, 9 and 10) to identify the best conditions for optimizing both the total
22 SCFAs and propionate content from CPS fermentation. Then, the optimal conditions were

23 applied in long term to a Sequencing Batch Fermentation Reactor where the highest
24 propionate production (100-120 mg COD/g TVS_{fed}·d) was obtained at 37°C and adjusting the
25 feeding pH at 8. This was attributed to the higher hydrolysis efficiency of the cellulosic
26 materials (up to 44%), which increased the selective growth of *Propionibacterium*
27 *acidopropionici* in the fermentation broth up to 34%. At the same time, around 88% of the
28 phosphorus released during the acidogenic fermentation was recovered as much as 0.15 kg of
29 struvite per capita·year. Finally, the potential market value was preliminary estimated for the
30 recovered materials that can triple over the conventional scenario of biogas recovery in
31 existing municipal wastewater treatment plants.

32
33 **Keywords:** cellulosic primary sludge; acidogenic fermentation; propionate; resource
34 recovery; struvite

35 **Highlights**

- 36 • Separation and fermentation of cellulosic primary sludge enable wastewater-based
37 biorefinery
- 38 • Initial pH (8) and T (37 °C) selected *Propionibacterium acidopropionici* which
39 optimized the propionate recovery
- 40 • Struvite recovery from CPS fermentation liquid best integrates the biorefinery concept
- 41 • The CPS-based biorefinery could triple the economic value of wastewater

42 **1. Introduction**

43 Primary sludge (PS) usually contains a large quantity of biodegradable organic compounds
44 such proteins, carbohydrates, cellulose and other organic materials. Among them, cellulose
45 represents approximately 30-50% of the influent suspended solids in wastewater treatment

46 plants (WWTPs) of Western European countries (STOWA report, 2012) where toilet paper is
47 flushed into the sewers system. In these countries, the average per capita consumption of
48 toilet paper was estimated around 15 kg per year which is 3 times more than the global
49 average consumption (4.4 kg/capita·year) and 10 times more if compared with the
50 consumptions of developing countries (Pulp and Paper Industry Intelligence, 2011). Due to
51 the flushing of toilet paper in public sewers, in Western European countries cellulose usually
52 enters municipal wastewater treatment plants (WWTPs) and is only partially degraded and
53 valorised. However, flushing toilet paper may probably be considered more environmentally
54 friendly practice compared to disposal in toilet trash and following transportation to landfills
55 or incinerators, that is implemented in countries where sewers infrastructure can have
56 clogging problems (Genty et al. 2013).

57 Usually, the rate-limiting step of cellulose degradation is the hydrolysis process (Noike et al.,
58 1985), which makes difficult its degradation during the conventional biological treatments in
59 WWTPs. Verachtert et al., (1982) reports that 60% of the cellulosic material is degraded
60 during 4-5 weeks of aerobic conditions, while 40% persists undegraded in the excess sludge.
61 However, if anaerobic digestion of excess sludge is accomplished, additional 50% of the
62 present cellulose could be degraded. Ruiken et al, 2014 carried out batch experiments to
63 investigate the mechanism of toilet paper under anaerobic conditions. The authors found
64 100% of removal after 8 days at 30°C of temperature, confirming that the cellulose
65 degradation is indeed a slow process. On the other hand, the presence of cellulose in activated
66 sludge and digested sludge calls for more in-depth studies on the conversion processes of
67 these fibres (Ruiken et al., 2014; Rusten et al., 2006). When properly separated and refined,
68 the cellulose can be used as raw material to make paper products or adhesion binders for
69 asphalts (STOWA report 2012; Godow et al., 2013). In addition, cellulose can be used to

70 produce valuable chemicals or biofuels, such as short-chain fatty acids (SCFAs), poly lactic
71 acid, bioethanol (Van der Hoek et al., 2015; Honda et al., 2002). The recovery of cellulosic
72 primary sludge (CPS) in a water resource recovery facility (WRRF) can be performed by fine-
73 mesh sieves (<500 μm) and the resulting primary sludge where the fraction cellulose achieves
74 79% of the total mass and 84% of the organic mass (STOWA report 2012; Ruiken et al.,
75 2014; Rusten et al., 2006). Currently, only few studies have investigated the best pathways to
76 valorise the CPS (Ruiken et al., 2014; Honda et al., 2002; Ghasimi et al., 2016), while
77 recovery of propionate-rich SCFAs have never been studied within a wider biorefinery
78 concept. In this regard, carbon upgrading to SCFAs, mainly acetate, propionate and butyrate,
79 etc, is a cost-effective strategy to produce intermediates which can be processed to
80 (bio)products with higher potential market value than methane (CH_4) from biogas
81 (Kleerebezem et al., 2015; Holtzapple et al., 2009). Moreover, recent studies suggest that
82 propionate can best enhance the biological phosphorus removal (BPR) processes in biological
83 nutrients removal systems (Chen et al., 2004; Oehmen et al., 2006). On the other hand, higher
84 propionate/acetate ratio promotes the selective growth of polyphosphate accumulating
85 organisms compared to the glycogen accumulating organisms in enhanced biological
86 phosphorus removal systems (Oehmen et al., 2006). In addition, SCFAs with higher
87 propionate/acetate ratio promote the production of co-polymers characterised by low stiffness
88 and brittleness, higher flexibility (higher elongation to break), and higher tensile strength and
89 toughness (Laycock et al., 2014; Frison et al., 2015). Consequently, the selective production
90 of SCFA from sewage sludge under optimized acidogenic fermenting conditions have become
91 an emerging research field that enables wastewater-based biorefineries (Lee et al., 2014;
92 Basset et al., 2016).

93 Other authors (Zurzolo et al., 2016) studied the SCFAs production from the fermentation of
94 conventional primary and secondary sludge, while the potential of SCFAs production and
95 nutrients recovery (e.g. struvite) from CPS is still unknown. In this regard, the rates and the
96 yields of CPS fermentation could be influenced by key operating parameters such as pH,
97 sludge retention time (SRT) and temperature. pH affects the hydrolysis and the subsequent
98 acidification step during the fermentation. Alkaline conditions (pH >9) promotes the
99 fermentation of primary sludge and inhibits the methanogenic activity, to achieve higher
100 conversion to SCFAs (Wu et al., 2010). On the other hand, it is reported that the optimum pH
101 range for the hydrolysis and acidogenesis of cellulose is between 5.6 and 7.3 (Hu et al., 2004).

102 The fermentation of sewage sludge is usually performed at mesophilic (30-40 °C) or moderate
103 thermophilic (50-55 °C) conditions. Thermophilic condition may increase the substrate
104 degradation rate, but this is unfavourable for the both energy balance and for the process
105 stability. Therefore, mesophilic conditions are still recommended to achieve a robust and
106 stable sludge fermentation (Yu et al., 2002). However, the combined effect of operating
107 temperature, pH and SRT on the production of SCFAs from CPS fermentation is still unknown.

108 In addition, the fermentation of sewage sludge involves relevant orthophosphate release in the
109 liquid phase, which can be effectively recovered through struvite crystallization (Tong et al.,
110 2009; Zhang et al., 2009).

111 In this paper, the optimization of the SCFAs production was explored through the
112 fermentation of CPS at different temperature (37, 55 and 70 °C) and initial sludge pH
113 (uncontrolled pH, 8, 9 and 10) to maximize: (1) the production of propionate (Pr) in SCFAs,
114 (2) the recovery of phosphorus (PO₄-P) as struvite from the fermentation liquid; (3) the final
115 biogas production from the fermentation solid. The resulting optimized parameters were then
116 used to set-up and study the long-term operation of a Sequencing Batch Fermentation Reactor

117 (SBFR), that provided the results for the forthcoming scale-up for WRRFs. Finally, the
118 market added value of the recovered materials from CPS fermentation was estimated based on
119 the experimental results. Based on the rates and mass flows obtained in the SBFR, the scale-
120 up of this scheme will be integrated in the real wastewater treatment plant of Carbonera
121 (Treviso) within the European Horizon2020 Innovation Action “SMART-Plant”.

122

123

124 **2. Material and Methods**

125 *2.1 Source and physicochemical characteristics of the CPS*

126 Real and raw thickened PS was collected once per week and for 3 months during spring
127 season and dry weather from the municipal WWTP of Verona (Veneto Region, Italy).

128 Primary sludge was settled in primary clarifiers after the removal of grit, sand particles and oil
129 from the influent wastewater. Then, the PS was thickened up to around 4% total solids (TS)

130 using gravity belt thickening (Klein Technical Solutions, Germany). After sampling the PS

131 was mixed and homogenized with toilet paper obtaining a total concentration of cellulose

132 around 70-75% in agreement with other literature studies (Ruiken et al., 2014). Before the

133 preparation of CPS, the toilet paper was soaked in wastewater for 4 h to achieve a CPS with

134 similar characteristics to those found in WWTPs. The main characteristics of CPS were as

135 follows: total solids (TS) 56.0 ± 17.2 g TS/L, volatile solids (TVS) 48.2 ± 14.0 g TVS/L, pH

136 of 6.3 ± 0.1 , total chemical oxygen demand (tCOD) and soluble chemical oxygen demand

137 (sCOD) concentrations of 949 ± 156 mg COD/g TVS and 48 ± 26 mg COD/g TVS,

138 respectively, ammonium concentration ($\text{NH}_4\text{-N}$) 1.7 ± 0.4 mg N/g TVS and orthophosphate

139 concentration 0.5 ± 0.1 mg $\text{PO}_4\text{-P}$ /g TVS. The concentration of total SCFAs detected in the

140 CPS was 26.2 ± 9.8 mg COD/g TVS.

141

142 *2.2 Operating temperatures and pH of the CPS fermentation*

143 In order to acquire full understanding of inputs and outputs being investigated, the complete
144 matrix of batch fermentation experiments of CPS was performed at: (a) different
145 temperatures, 37 ± 1 °C (mesophilic), 55 ± 1 °C (thermophilic) and 70 ± 1 °C
146 (hyperthermophilic); (b) different initial value of pH (8, 9 and 10) to investigate the effect of
147 these operating parameters on the SCFAs production and composition. The batch
148 fermentation experiments were performed in duplicates by using 1 L glass bottles, with a
149 working volume of 0.6 L. The initial pH was adjusted using sodium hydroxide (NaOH). In
150 addition, a batch experiment with uncontrolled pH was used as the reference fermenting
151 conditions. The batch assays were kept at controlled temperature for 16 days, while pH was
152 not controlled during the fermentation tests. The reactors were sealed with rubber stopper and
153 opened for only approximately 1 min for sampling and to measure the pH using a pH sensor
154 (Eutech pH 700). The samples were centrifuged, filtered through membrane filters (Whatman,
155 $0.45 \mu\text{m}$), then analysed for $\text{PO}_4\text{-P}$, $\text{NH}_4\text{-N}$, sCOD concentrations and SCFAs concentration
156 and composition. Total solids (TS) and volatile solids (TVS) were determined at the
157 beginning and end of the fermentation experiments.

158 The actual production of SCFAs was always calculated subtracting the initial SCFAs
159 concentration of the raw sludge. The yield of SCFAs production was expressed as mg COD/L
160 of SCFA per g TVS/L in the feed sludge ($\text{mg COD/g TVS}_{\text{fed}}$). Similarly, the released $\text{PO}_4\text{-P}$
161 was determined as $\text{mg PO}_4\text{-P/gTVS}_{\text{fed}}$.

162

163 *2.3 Optimization of the CPS fermentation to enhance propionate production*

164 Based on the full set of batch experiments the response surface methodology (RSM) was
165 applied to further advance the propionate production based on the fermenting temperature and
166 initial pH value. The regression model used is shown in Equation 1 and the target responses
167 were the production of SCFAs (Y_{SCFAs}) and the content of propionate over those SCFAs
168 (%Pr).

$$169 \quad Y(z) = \beta_0 + \beta_1 x_1 + \beta_2 x_2 + \beta_{1,2} x_1 x_2 + \beta_{1,1} x_1^2 + \beta_{2,2} x_2^2 \quad (\text{Eq. 1})$$

170 where $Y(z)$ is the response variable (i.e. Y_{SCFAs} (mg COD/g TVS_{fed}) or %Pr (gCOD/gCOD x
171 100), x_1 is the initial pH and x_2 is the temperature (°C), β_0 is the model constant, β_1 and β_2
172 are linear coefficients, $\beta_{1,2}$ is the cross-product coefficient, and $\beta_{1,1}$ and $\beta_{2,2}$ are the quadratic
173 coefficients.

174 Finally, a statistical analysis was carried out by means of the analysis of variance (ANOVA)
175 to test the significance of predicted and experimental results, under a significance level of
176 0.05 (p). The regression model and the statistical analyses were performed using the software
177 R 3.2.3 (The R Foundation for Statistical Computing).

178

179 *2.4 Sequencing Batch Fermentation Reactor*

180 A sequencing batch fermentation reactor (SBFR) with a reaction volume of 4 L was operated
181 at 37 ± 1 °C by a thermostatic bath, while the HRT was kept constant at 4 days by the daily
182 exchange of 25% of the reactor volume between fermented and fresh CPS. The SBFR was
183 equipped with a blade stirrer installed in the bottom. In this work, two representative
184 experimental periods were carried out (period 1 and 2) where the steady-state conditions were
185 considered when the calculated Relative Standard Deviation (RSD) of the propionate

186 production was below 10% for at least 3 times the HRT of the SBFR (Ghasem et al., 2008).

187 The Equation 2 reports how the RSD was calculated:

$$\text{RSD (\%)} = \frac{\sigma (\text{Pr production})}{\mu (\text{Pr production})} \times 100 \text{ (Equation 2)}$$

188 where σ (Pr production) is the standard deviation of the propionate production found during
189 the period, while μ (Pr production) is the average of the propionate production found during
190 the period.

191 The period 1 (0-18 days) was operated to determine the yield of Pr production without any
192 adjustment of pH in the fed CPS. In the period 2 (19-96 days), every day the pH of the fresh
193 CPS was adjusted to 8 before the feeding of the reactor. Steady-state conditions were
194 achieved between days 4 and 16 for period 1, while during days 36-96 for period 2.

195 Samples were periodically taken from the effluent of the SBFR and analysed for $\text{PO}_4\text{-P}$, $\text{NH}_4\text{-}$
196 N , SCFAs, chemical oxygen demand (total COD, soluble COD), total solids (TS), volatile
197 solids (TVS) and pH. The composition of the SCFAs (i.e. acetate, propionate) were also
198 investigated to determine the propionate/acetate ratio ($\text{g COD}_{\text{propionate}}/\text{g COD}_{\text{acetate}}$) as
199 monitoring parameter during the experiment. Propionate/acetate between 0.25-0.75 g
200 $\text{COD}_{\text{propionate}}/\text{g COD}_{\text{acetate}}$ is considered the optimal biological phosphorus removal processes
201 (Broughton et al., 2008; Yuan et al., 2012). On the other hand, higher $\text{COD}_{\text{propionate}}/\text{g}$
202 $\text{COD}_{\text{acetate}}$ ratios promote the production of polyhydroxyvalerate (PHV) instead of
203 polyhydroxybutyrate (PHB) improving the mechanical and physical properties of the
204 biologically recovered PHAs (Jiang and Chen, 2009).

205 Twice per week the “cellulosic materials” in the influent and effluent from the SBRF were
206 quantified as the volatile fraction at 550°C of the solids recovered and washed after the
207 sieving at mesh 54 μm . During the periods 1 and 2, samples of biomass from the SBFR were

208 taken and characterized by the FISH quantification of the *Propionibacterium acidopropionici*
209 using the Apr820 and DAPI probes, following the methodology described Nielsen et al.
210 (2009). Forty images of each sample were taken using a fluorescence microscope (Leica
211 DM2500) and then analysed with the Image J software.

212 During the steady-state conditions of the SBFR, struvite recovery tests were performed from
213 the CPS fermentation liquid. The fermentation liquid from the supernatant was obtained after
214 the centrifugation at 4000 rpm for 10 minutes of the effluent from the SBFR. The experiments
215 started with the addition of 5 g/L of seed struvite crystals and magnesium hydroxide
216 ($\text{Mg}(\text{OH})_2$) according to a PO_4^{3-} : Mg^{2+} molar ratio of 1:1.5, and adjusting the initial pH at 8.5
217 with NaOH (0.1 M). Samples were collected at 5 min, 10 min, 15 min, 30 min and 60 min,
218 filtered through cellulose membrane filters (Munktell Ahlstrom) and analysed to determine
219 their $\text{PO}_4\text{-P}$ concentration. After the experiments, the precipitated solids were washed with
220 distilled water to remove impurities and soluble salts. The recovered solids were dried at 45
221 °C for 24 h to avoid thermal decomposition (Bhuiyan et al., 2008). The crystals produced
222 were analysed according to Fattah et al. (2012) and the molar ratio between nitrogen and
223 phosphorus was used to confirm the struvite formation.

224

225 *2.5 Biochemical methane potential tests*

226 The BMP test were also investigated. The BMP test were carried out following the procedure
227 defined by Angelidaki et al., (2009) at 37 °C, while parallel tests were carried out using raw
228 primary sludge to compare biogas production and composition from CPS and PS BMP tests.
229 More details of this method are reported in Supporting Information.

230

231 2.6 Analytical methods

232 Soluble COD, TSS and TVS were measured according to Standards Methods (APHA-
233 AWWA-WPCF, 2012). $\text{NH}_4\text{-N}$ concentration was measured by an ion selective electrode
234 (Orion 9512). The concentration of SCFAs was determined by gas chromatography (Dionex
235 ICS-1100 with AS23 column). $\text{PO}_4\text{-P}$ concentration was measured by ion chromatography
236 (Dionex ICS-900 with AS14 column) and calibrated using a combined five anion standard
237 (Thermo Scientific™ Dionex™ Ion Standards).

238

239 3. Results and Discussion

240 3.1 SCFAs production and composition from the batch fermentation experiments

241 No relevant lag-phase was observed before the production of SCFAs started. The
242 concentration increased gradually until the peak and plateau values were reached (between the
243 9th and 13th day) at 37 and 55 °C (Figures S1, Supplementary Material). On the other hand,
244 the SCFAs production at 70 °C increased up to day 2 and then decreased (Figure S1 in the
245 Supplementary Material). The highest production yield of SCFA of 340.4 mg COD/g TVS_{fed}
246 was observed at 37 °C (Figure 1), while the lowest were 155.4 and 46.1 mg COD/g TVS_{fed}
247 observed at 55°C and 70°C respectively. Therefore, the temperature had a major effect on the
248 production of SCFAs. In particular, the effect the fermentation temperature had on the
249 hydrolyses and the acidogenic process can clearly be detected from the pH profile over the
250 batch experiments (SI Figure S2), since it was adjusted only at the beginning of the
251 experiments. Moreover, higher variation of pH resulted in higher SCFAs production, thus pH
252 seemed to act as a surrogate parameter for monitoring the fermentation process.

253

254 (Figure 1)

255

256 Although the increase of pH can positively influence the fermentation efficiencies (Wu et al.,
257 2010), in this study the highest SCFAs production were observed at pH 8, while at pH 9 and
258 10 the production of SCFAs were much lower at 37 °C (Figure 1). On the other hand, under
259 thermophilic and hyper-thermophilic conditions, fermentation pH was almost stable and the
260 production of SCFAs was comparable notwithstanding the initial pH condition.

261 Acetate and propionate were the most relevant SCFAs in the fermentation liquid for all the
262 fermentation experiments at 37 °C (acetate 55-80%, propionate 12-33%), with minor
263 concentration of mainly butyrate and n-valerate (around 6-9% and 3-4% respectively)
264 (Figures S3(a), S3(b), S3(c) in the Supporting Information). Propionate was mostly produced
265 under mesophilic condition in a range of percentage between 25-33%, while under
266 thermophilic and hyper thermophilic propionate contents below 20% were observed. The
267 fermentation temperature increase led to propionate/acetate ratio decrease: the higher ratio of
268 0.6 gCOD/gCOD was observed at 37 °C and initial pH of 8.

269

270 *3.2 Optimal key operating parameters for CPS fermentation*

271 The 3D response surface methodology (RSM) was applied to optimize both the total SCFAs
272 production and the propionate percentage based on the initial pH and the fermentation
273 temperature. The results of the regression model indicated that the increase of the
274 fermenting temperature has a detrimental effect on total SCFAs and Pr productions than
275 initial pH fixed value (SI Figure S2a and 2b). Figure 2a and 2b shows that SCFAs and Pr
276 percentage productions decrease with the increase of fermentation temperature, obtaining

277 higher SCFAs productions from CPS fermentation at 37 °C. So, the response surface plot
278 indicates that the most favourable operating conditions to maximize SCFAs production were
279 37 °C of temperature and initial pH of 8 (Figure 2a), while the highest percentage of
280 propionate can be obtained at lower initial pH ($7.5 < \text{pH} < 8.0$) and a temperature of 37 °C.
281 Under these operating conditions the production of SCFAs and propionate percentage were
282 281.5 mg COD/g TVS and 30.3%, respectively (Figure 2a and 2b).

283

284 (Figure 2a and 2b)

285

286 The statistical significance of total SCFAs production and propionate percentage models were
287 evaluated by ANOVA (Table 1).

288

289 (Table 1)

290

291 The combination of low p value and high R^2 indicated that the model explains a lot of
292 variation within the data and is significant. The models of F-value showed a low p-value
293 (< 0.016), which implied that both models were significantly affected by temperature and
294 initial pH, and able to estimate total SCFAs production and %Pr. The predicted high R^2
295 (between 0.7 and 0.9) indicated both models were sufficiently high to show the significance
296 of the fit of the models (Table 1). Table 2 reported the coefficients of the quadratic models for
297 the responses of the Y_{SCFA} and %Pr. For %Pr model, pH, temp and the interaction effect $\beta_{1,2}$
298 and $\beta_{2,2}$ were not significant ($p > 0.05$), while only $\beta_{1,1}$ was found significant ($p < 0.05$). For

299 SCFAs production model, all the terms were found not significant ($p>0.05$) meaning that the
300 effect on response is considerable.

301 (Table 2)

302 Figure 3 shows the comparison between the predicted and experimental values for the total
303 SCFAs and the percent of propionate production. The experimental results are consistent with
304 the regression model ($R^2 >0.86$) for both key parameters.

305

306 (Figure 3)

307

308 *3.3 Long-term SCFAs production and phosphorus recovery in the SBFR*

309 During the period 1 (0-18 days), the sCOD varied in the range 10-12 gCOD/L and lower
310 yields of Pr production were observed (around 60 mgCOD_{propionate}/gTVS_{fed}·d). In period 2, the
311 sCOD higher and more stable in the range 14-18 gCOD/L (see Figure S4, Supporting
312 Information). The increase of the sCOD was a result of the higher degradation of the
313 cellulosic materials observed in period 2 (Table 3).

314

315 (Table 3)

316 The effect of the influent pH was clearly observed by the productivity of the SCFAs, which
317 increased from 162.4±12.8 mg COD/g TVS_{fed}·d (period 1) to 253.8±26.1 mg COD/g
318 TVS_{fed}·d (period 2). Moreover, the increase of the pH influent had effect on the propionate
319 production, which gradually increased and reached a stable production of 100-120 mg COD/g
320 TVS_{fed}·d during days 36-96 of period 2 (Figure 4). The latter is higher than the batch
321 experiments, probably due to the speciation of the microbial community accomplished in the
322 SBFR. In fact, the percentage of propionate to total SCFAs was up to 46%. This corresponded

323 to propionate/acetate ratio of 0.9 gCOD/gCOD, which was higher than the period 1 (0.6
324 gCOD/gCOD).

325

326 Many authors reported that the fermentation of cellulosic compounds at relatively low pH,
327 lactic acid could be produced (Abdel-Rahman et al., 2013). Indeed, despite the initial pH of
328 the CPS in period 2 (19-96) was adjusted to 8, the average pH in the SBFR dropped to
329 5.1 ± 0.1 due to simultaneous production of the SCFAs and alkalinity consumption (Figure 4).
330 As a consequence, bacteria of the genus *Propionibacterium* may produce propionate from
331 lactate as the end-product of their anaerobic metabolism (Liu et al., 2012). In period 2, FISH
332 analyses (see Table S1 and Figure S6 in Supporting Information) confirmed selective growth
333 of *Propionibacterium acidopropionici* that were at 33.8%, more abundant than period 1
334 (24.5%). So, the beneficial speciation of the microbial community in the SBFR seems to be
335 related with the higher solubilization of the cellulosic materials achieved in period 2.

336

337 (Figure 4)

338

339 Furthermore, $\text{NH}_4\text{-N}$ and $\text{PO}_4\text{-P}$ were released in the CPS fermentation liquid of the SBFR as
340 much as 1.6 ± 0.5 mg P/g $\text{TVS}_{\text{fed}} \cdot \text{d}$ and 6.1 ± 1.4 mg N/g $\text{TVS}_{\text{fed}} \cdot \text{d}$. As a consequence, the
341 average concentrations of $\text{PO}_4\text{-P}$ and $\text{NH}_4\text{-N}$ in sludge fermentation liquid during steady
342 conditions were 130 ± 23 mg P/L and 430 ± 29 mg N/L, respectively. Due to the high P and N
343 content, the recovery of the released $\text{PO}_4\text{-P}$ from CPS fermentation by struvite crystallization
344 was examined and the average efficiency of phosphorus recovery was 88%.

345

346 *3.4 Specific per capita recovery of valuable resources from CPS and preliminary economics*

347 The selective production of mixture of SCFAs through acidogenic fermentation was
348 considered best available carbon source to enhance the nutrients removal in the mainstream or
349 in the sidestream (Frison et al., 2013, 2016). Recently, Longo et., 2017 considered this
350 practice an economic and environmentally friendly solution to reduce energy and chemical
351 consumption for the removal of nitrogen and phosphorus. Moreover, a number of
352 Horizon2020 Innovation Actions (such as SMART-Plant (www.smart-plant.eu) or INCOVER
353 (www.incover.eu) were evaluated and funded to demonstrate the technical, economic and
354 environmental long-term viability of further (bio)conversion of the SCFAs to biopolymers,
355 such as PHAs, that could enable the recovery of high added value products by minor
356 integration of existing WWTPs. Although the best technical and economical evaluation
357 should always be referred to single WWTP, the specific economic advantages of alternative
358 for CPS processing in comparison with the only biogas production is estimated below.

359 Every year, around 36-43 kg of COD are discharged in municipal wastewater by individuals
360 (capita) (adapted Metcalf and Eddy, 2014). The observed average removal efficiency of COD
361 by the sieving municipal wastewater is between 10-60% (Ruiken et, 2014) and 12-13 kg
362 COD/capita·year could be recovered as suspended solids from municipal wastewater. In this
363 study the BMP test (SI Figure S5) showed that CPS may produce up to 0.30-0.34 m³ CH₄/kg
364 COD_{fed} equivalent to 3.7-4.5 m³ CH₄/capita·year (Figure 5a), which is in agreement with
365 other studies (Ghasimi et al., 2016). Therefore, the convertible COD to CH₄ by anaerobic
366 digestion in a current WWTP (Table 4) would be around 11 kg COD/capita·year, which
367 represents around 25% of the total COD influent in a WWTP. In a CPS-based biorefinery
368 scenario (Table 4), considering the observed SCFAs production rate, around 3.0 kg
369 COD_{SCFA}/capita·year could be produced by the fermentation of CPS, where acetate and
370 propionate represent 1.30 and 1.17 kg COD_{SCFA}/capita·year, respectively. However, around

371 2.7-3.3 m³CH₄/capita·year of residual CH₄ could be further produced by the anaerobic
372 digestion of CPS after fermentation (Figure 5b) Moreover, during the fermentation of CPS,
373 nutrients are released and 88% of the phosphorus could be recovered in the form of struvite,
374 that amounts to 0.07-0.15 kg struvite/capita·year.

375 Assuming CH₄ a market price of 0.11 €/m³ (Energy Information Administration, 2017), the
376 best valorisation of CH₄ from CPS can be as high as 0.46 €/capita·year. On the other hand,
377 better value can derive from valorising CPS first to the suitable mix of SCFAs (mainly acetate
378 and propionate) and struvite from the fermentation liquid, while CH₄ can be recovered after
379 digestion of fermentation solids. Although the market price of the recovered materials is very
380 volatile and often unknown because of the variable purity and quality, according to a recent
381 review, acetate and propionate price can be as high as 0.45 and 1.01 €/kg respectively (Global
382 Chemical Price, 2017; ICIS, 2017), while struvite can be sold up to 0.76 €/kg (Molinos-
383 Senante et al., 2011; P-REX report, 2014). Therefore, the SCFAs and struvite route before the
384 bio-methanization can increase the market value potential of CPS up to 1.55-1.95
385 €/capita·year (Table 4).

386 (Table 4)

387 (Figure 5)

388

389 4. Conclusions

390 This paper studied the maximum potential recovery of SCFAs, particularly propionate,
391 struvite and CH₄ from the CPS. Based on the results of the RSM, the optimal production of
392 propionate is obtained by the fermentation of CPS at mesophilic conditions (37 °C) and at
393 initial pH between 7.5-8. By the long-term operation of a SBFR the observed production of
394 propionate in the fermentation liquid was 100-120 mg COD/g TVS_{fed}·d, with a

395 propionate/acetate ratio of 0.9 g COD/g COD. Best performances in the SBFR may be
396 attributed to the observed enhanced growth of *Propionibacterium acidopropionici*. At the
397 same time, 88% of the phosphate released in the fermentation liquid can be recovered as
398 struvite. From a techno-economic point of view, the integration of the wastewater dynamic
399 sieving to recover CPS in a WWTP may make existing units (e.g., gravity sludge thickener)
400 redundant and available to be revamped to controlled fermenter to recover optimal mixture of
401 SCFAs. In addition, the recovery of SCFAs and struvite before the bio-methanization can
402 increase the market value potential of CPS up to 1.55-1.95 €/capita·year.

403

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410

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- 1 Table 1. ANOVA results of the response surface quadratic model of the production of SCFAs
2 and propionate composition obtained from the fermentation of CPS.

Model	Std. Dev.	R ²	Adj. R ²	F-value	p-value
tSCFAs	9.2	0.92	0.8485	13.3	3.4 x10 ⁻⁴
%Pr	3.9	0.86	0.7392	7.2	1.6x10 ⁻³

3

- 1 Table 2. The coefficient of the quadratic models for the responses of SCFAs production and
 2 propionate percentage via analysis of variance (ANOVA).

Y, SCFAs (mg COD/gTVS_{fed})				
Coeficient	Estimate	Std. Error	t-value	p-value
β_0 (Intercept)	-24.6	599.1	-0.041	0.969
β_1 (pH)	173.7	124.0	1.400	0.211
β_2 (temp)	-13.0	10.8	-1.199	0.276
$\beta_{1,1}$ (pH ²)	-11.5	7.4	-1.565	0.169
$\beta_{2,2}$ (temp ²)	0.037	0.089	0.416	0.692
$\beta_{1,2}$ (pHxtemp)	0.3	0.6	0.453	0.666
Y, Pr (%)				
Coeficient	Estimate	Std. Error	t-value	p-value
β_0 (Intercept)	-21.8	59.9	-0.364	0.728
β_1 (pH)	30.2	12.4	2.432	0.051
β_2 (temp)	-2.4	1.1	-2.201	0.070
$\beta_{1,1}$ (pH ²)	-2.2	0.7	-2.991	0.024
$\beta_{2,2}$ (temp ²)	0.013	0.009	1.474	0.191
$\beta_{1,2}$ (pHxtemp)	0.1	0.1	1.184	0.281

Table 3: Degradation of the cellulosic materials observed during period 1 and period 2

Cellulosic materials	Unit	Period 1	Period 2
Influent	g/L	34.6±1.6	34.0±2.5
Effluent	g/L	23.5±1.2	17.8±2.7
% Degradation	%	32%±3%	44%±5%

Table 4. Market value of recovered materials from the fermentation of CPS (*residual methane production by anaerobic digestion of CPS fermentation solids)

Resource	Unit	Current WWTP		CPS-based biorefinery	
		Yield of Recovery	Revenue €/person year	Yield of recovery	Revenue €/capita year
Methane	m ³ CH ₄ /capita·year	3.7-4.5	0.41-0.49	2.7-3.3*	0.31-0.38
Acetate	kgCOD/capita·year	-	-	1.2-1.4	0.49-0.60
Propionate	kgCOD/capita·year	-	-	1.0-1.3	0.70-0.86
Struvite	kg/capita·year	-	-	0.07-0.15	0.05-0.11
Estimated Market Value	€/capita year	-	0.41-0.49	-	1.55-1.95

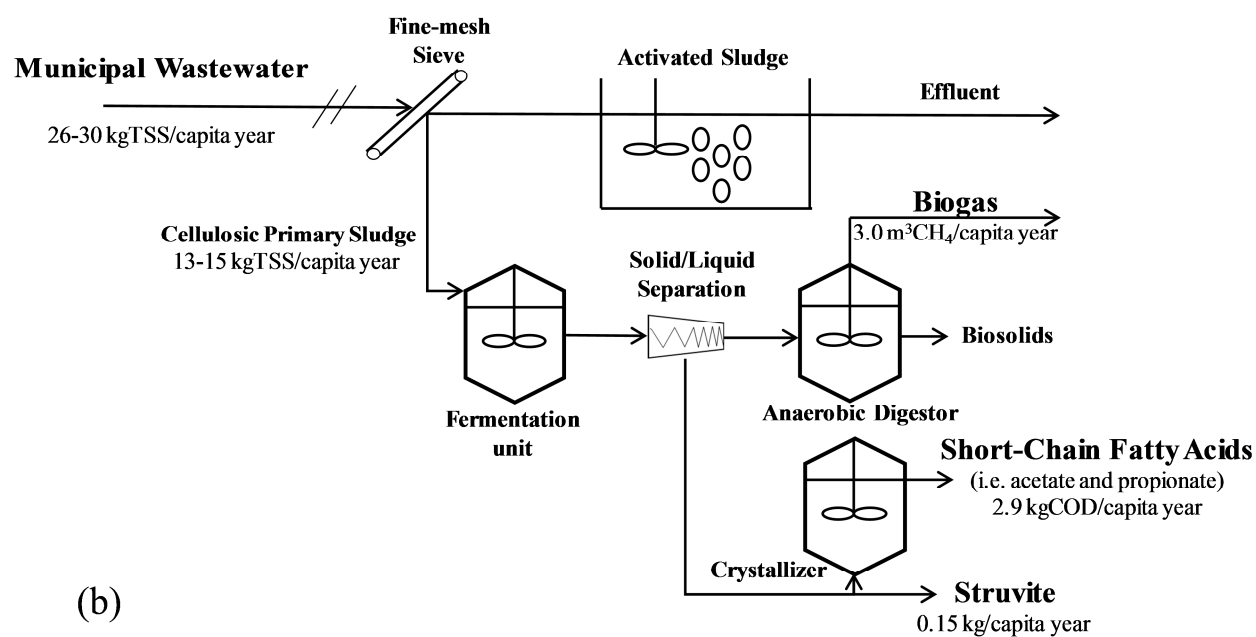
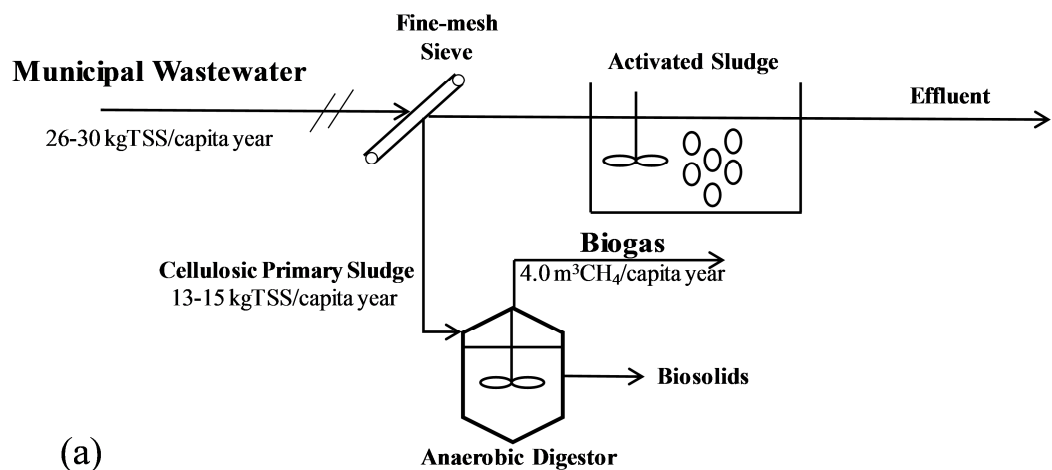


Figure 5. (a) Scheme for CPS valorization in current WWTPs; (b) Scheme for CPS valorization in a CPS-based biorefinery WWTPs.

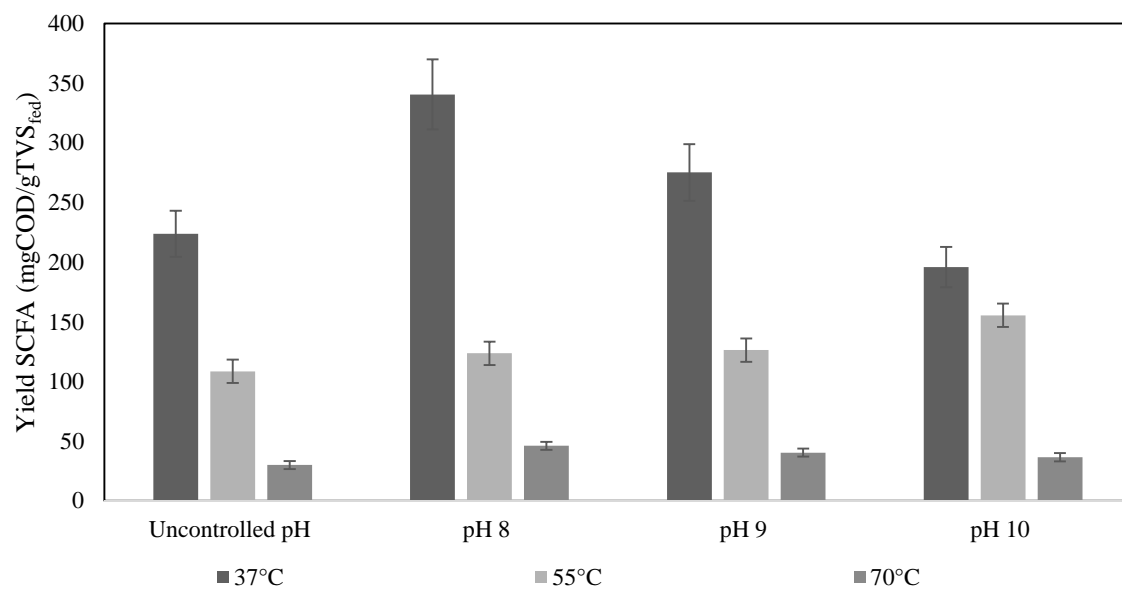


Figure 1. Optimal production of SCFAs obtained from the batch experiments.

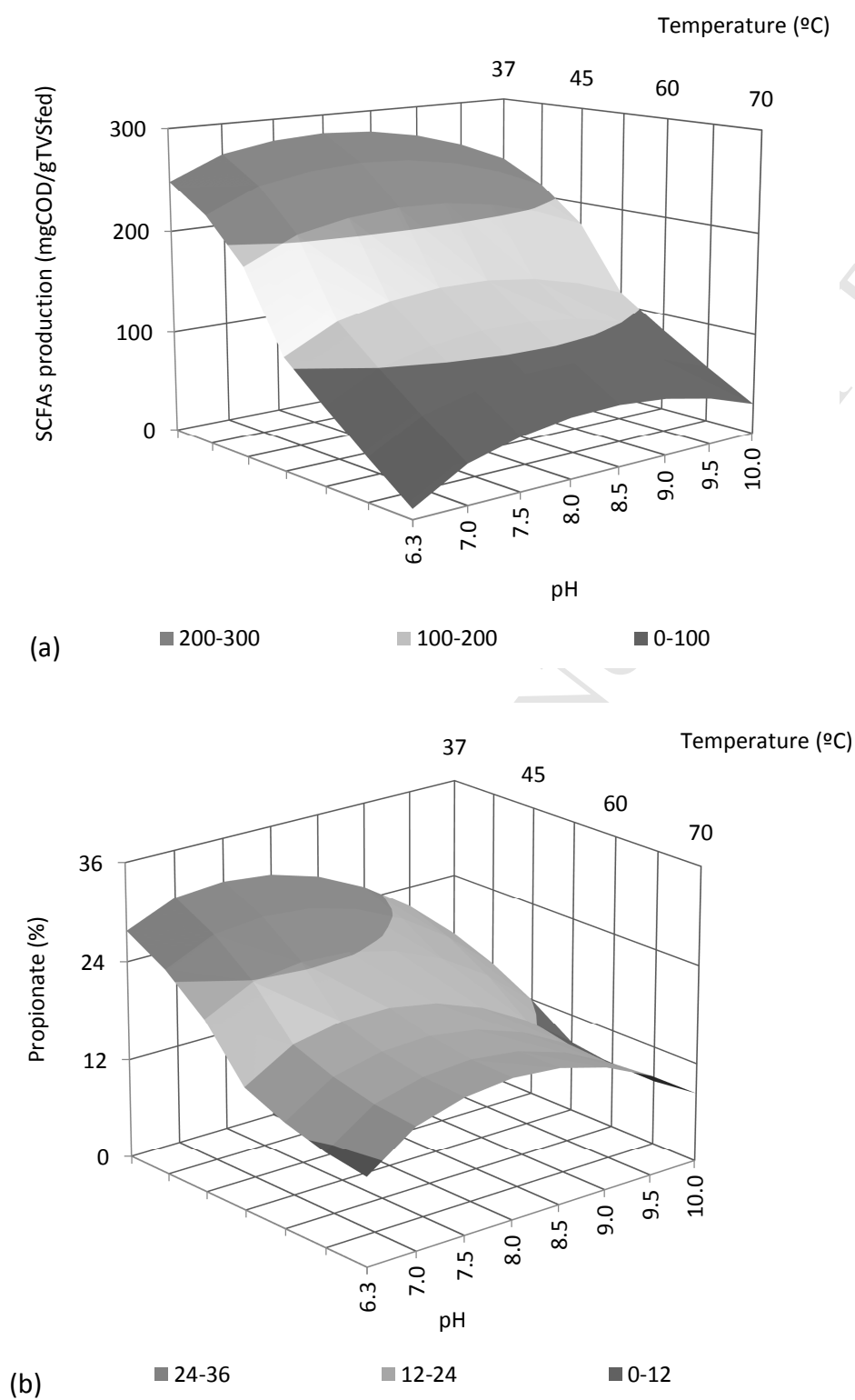
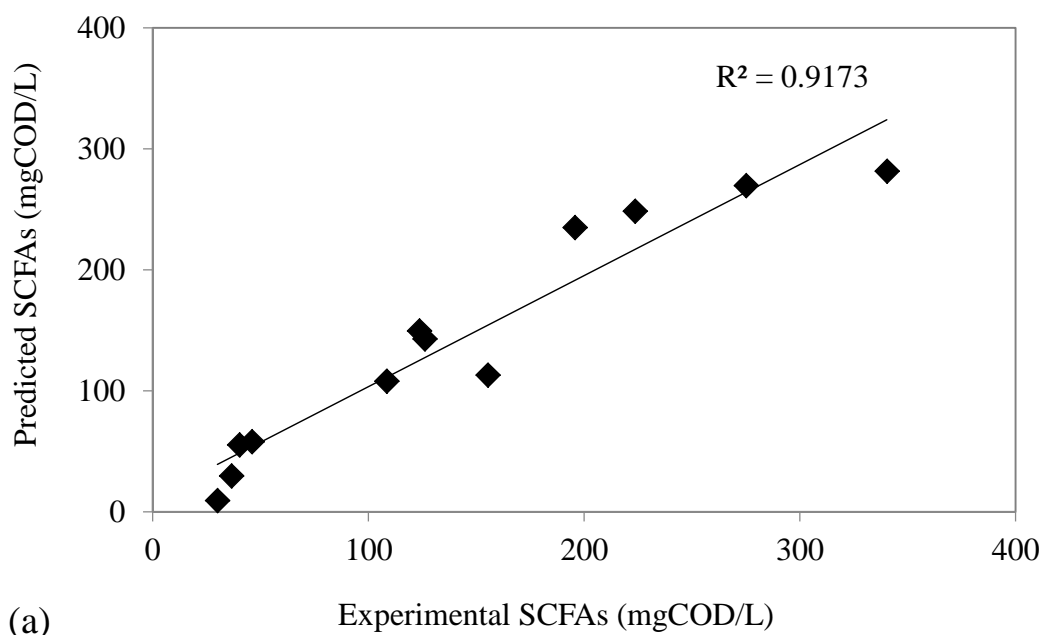
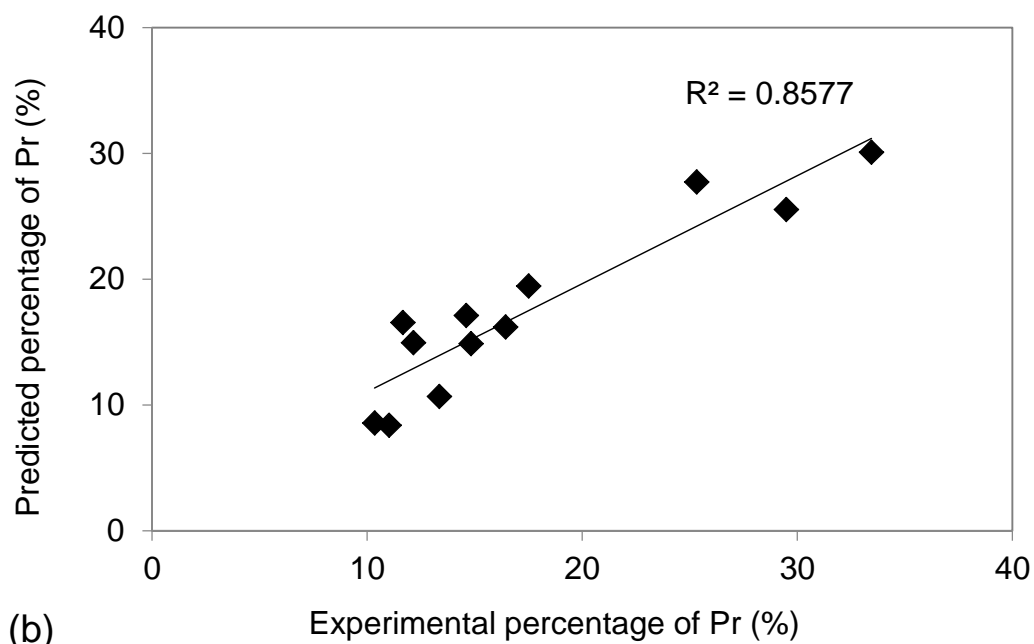


Figure 2. 3D surface plot response from the quadratic model for: (a) total SCFAs production and (b) propionate percentage.



(a)



(b)

Figure 3. Comparison between the experimental data and predicted values for SCFAs production (a) and percentage of propionate (b).

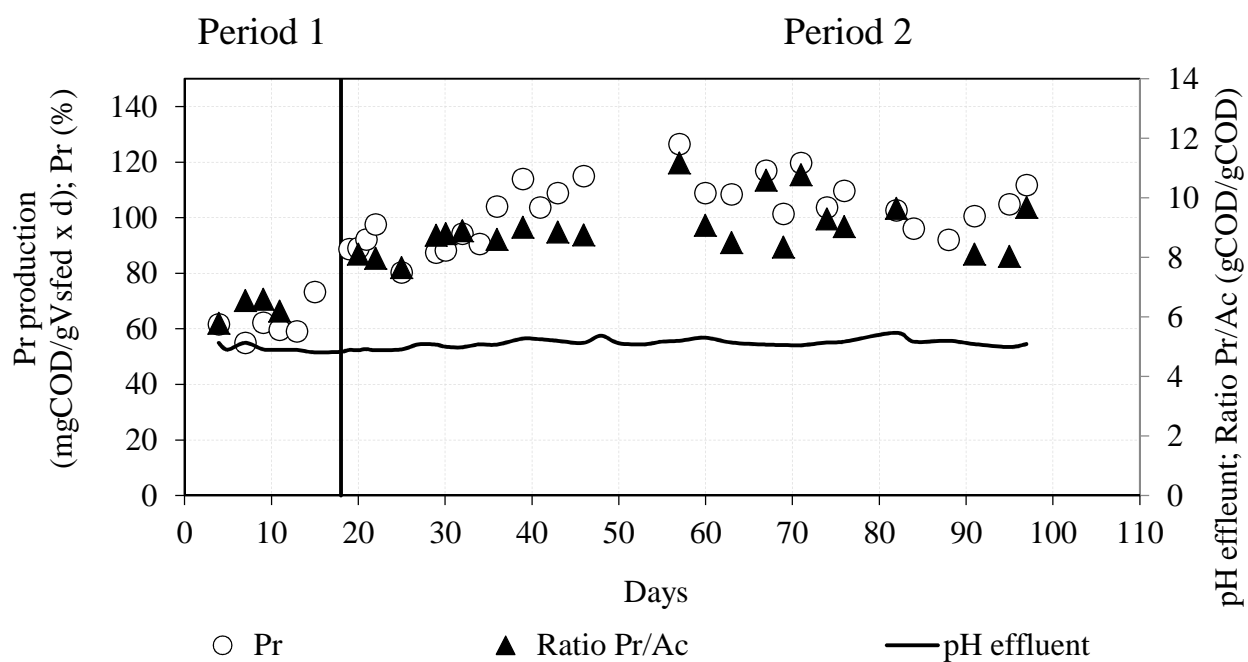


Figure 4. Profile of propionate production and operating pH during SBFR operation.

1 Highlights

- 2 • Separation and fermentation of cellulosic primary sludge enable wastewater-based
3 biorefinery
- 4 • Initial pH (8) and T (37°C) selected *Propionibacterium acidopropionici* which optimized
5 the propionate recovery
- 6 • Struvite recovery from CPS fermentation liquid best integrates the biorefinery concept
- 7 • The CPS-based biorefinery could triple the economic value of wastewater